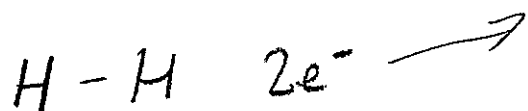


CP Example Of Work For Lewis Dot Structures (LDS)

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Below is the work expected when doing Lewis Dot Structures for covalent bonded substance (Molecular compounds, MC, or Polyatomic ions).

1. H_2 Total #ve⁻ : $2(1e^-) = 2e^-$

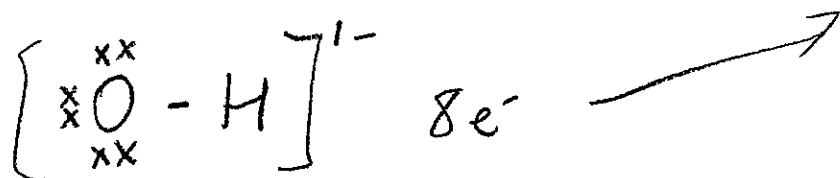


- No central atom (so don't need to tell me how many lone pair (LP) on central atom)

- 1 single bond

2. $O_1H_1^{1-}$ (called hydroxide ion)

Total #ve⁻ : $1(6e^-) + 1(1e^-) + 1e^- = 8e^-$



- No central atom

- 1 single bond

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