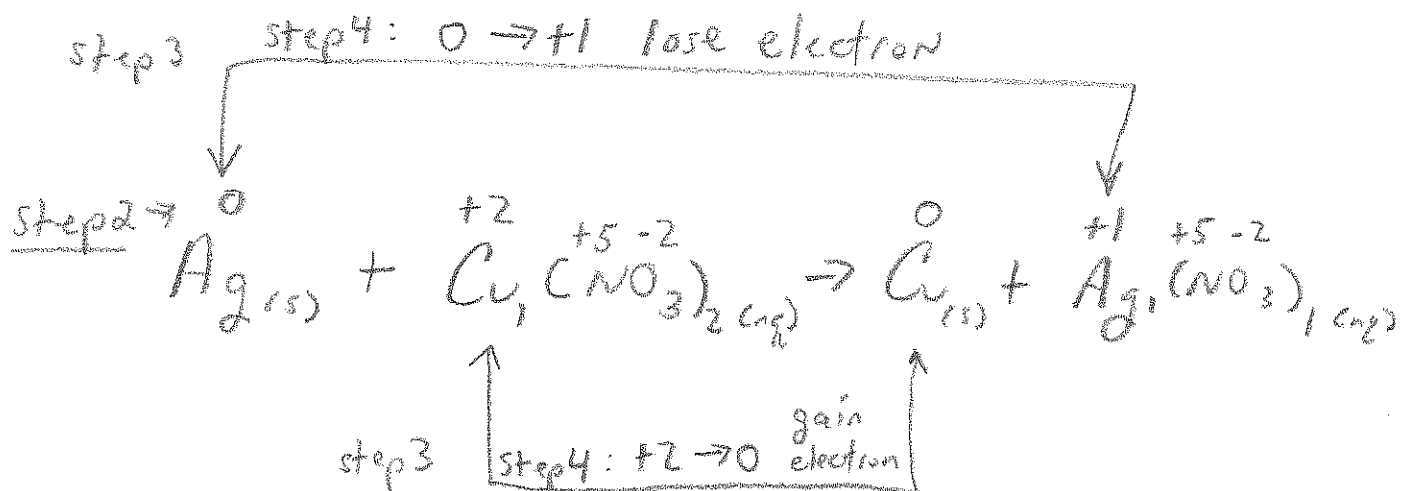
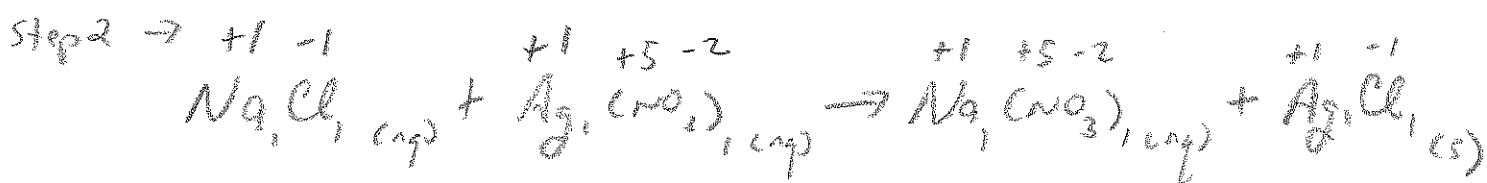


Example of work for Determine Redox Reactions



Step 5 LEO the Lwin goes GER

step 6 $\overset{+2}{\text{Cu}}$ is reduced Ag is reducing agent
 Ag^0 is oxidized $\text{Cu}(\text{NO}_3)_2$ is oxidizing agent



Since no oxidation number changed, this is not a Redox reaction.